

# Superior CO<sub>2</sub> Electrochemical Reduction Performance of Cu@AuCu Catalysts

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## Abstract

The electrochemical CO<sub>2</sub> reduction reaction (CO<sub>2</sub>RR) represents a viable alternative to help close the anthropogenic carbon cycle and convert intermittent electricity from renewable energy sources to chemical energy in the form of value-added chemicals. Development of economic catalysts possessing both high faradaic efficiency (FE) and mass activity toward CO<sub>2</sub>RR is critical in accelerating CO<sub>2</sub> utilization technology. In this study, a composite structured Au-Cu catalyst where an alloyed AuCu shell caps on a Cu core (Cu@AuCu) is developed and evaluated for CO<sub>2</sub>-to-CO electrochemical conversion. X-ray absorption spectroscopy (XAS) was performed for deeper structural insights of the Au-Cu catalysts, showing that the  $\chi_{\text{Au-Cu}}$  value, the hetero-atomic intermix (Cu) around Au atoms dramatically dropped down from 52% to 33%, as the bulk composition varied from Au<sub>1</sub>-Cu<sub>9</sub> to Au<sub>3</sub>-Cu<sub>7</sub>. Overall, Au atoms distribute more uniformly in the shell region at a lower Au content and are able to bond with more Cu atoms around. However, the coordination probability of Au-Au bonds increases at a higher Au content since homogeneity Au atoms prefer to gather, thus, leading to the reduced intermix of Cu around Au atoms. Additionally, the Cu@AuCu catalyst shows a compositional-dependent volcano-plot for selective CO production, which is balanced by the free energy for COOH\* formation and CO\* desorption. The optimized Au<sub>2</sub>-Cu<sub>8</sub> catalyst achieves a FE<sub>CO</sub> as high as 94% and a superior CO mass activity of 439 mA/mg<sub>Au</sub> at -0.8 V, particularly, only at an Au content of 17%. Insights gained in this study will not only shed new lights on the structure-property relationship of Au-Cu catalysts, but also benefit the design principles for high-performance CO<sub>2</sub>RR catalysts at high efficiency of noble metal utilization.

**Keywords** - electrochemical CO<sub>2</sub> reduction reaction, core/shell, CuAu, X-ray absorption spectroscopy, faradaic efficiency

## References

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